

Harnessing plant–microorganism interactions for nano-bioremediation of heavy metals: Cutting-edge advances and mechanisms

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ABSTRACT

Nano-bioremediation, an emerging eco-friendly strategy that integrates nanotechnology and biological processes to mitigate the contamination of heavy metals from the environment. To explore synergistic interactions between plants and microorganisms, focusing on their potential role in enhancing nano-bioremediation are highly demandable. This study focuses several key mechanisms including biosorption, bioaccumulation, biomineralization, and enzymatic reduction, the coordination of microorganisms and plants in tolerating and transforming toxic heavy metals into less toxic forms. The potential role of microorganism-assisted nanomaterials, including nano-biosorbents and nano-catalysts in phyto- and eco-environments were updated. This review also highlights recent studies on the significance of plant-microbe systems and nanomaterials in heavy metal remediation, challenges such as microbial survival, scalability, and ecological impacts were addressed, alongside potential solutions. Finally, this critical review provides new insights into harnessing plant–microorganism interactions for nano-bioremediation, presenting an eco-friendly approach to address global heavy metal pollution, and it shows a sustainable way of clean environment.

INTRODUCTION

Bioremediation (BR) is one of the most efficient methods for removal of heavy-metal from the contaminated soil and groundwater. This procedure is less harmful to the environment and more economical than conventional chemical and physical techniques, which are very costly and inefficient at low metal concentrations and result in large volumes of hazardous sludge [1,2]. The ability of microorganisms to breakdown contaminants is dependent on environmental conditions for growth and metabolism, which include favorable temperature, pH, and moisture [3,4]. Microorganisms are crucial for the removal of contaminants from soil, water, and sediments because of their benefits over alternative remediation techniques. These approaches of HMs removal are eco-friendly and cost effective. Additionally, they aid in the restoration of the natural ecosystem by preventing ongoing contamination [5]. Also, the contamination of heavy metals (HMs) has become a severe hazard to the ecosystem [6]. In the industrial sector, heavy metal is a profitable industry. Nevertheless, it is also a major environmental problem everywhere [7,8]. Also, the environment contains natural, agricultural, solid waste, inland effluents, and air sources, for the additional heavy metal sources. Large portions of the earth have been polluted by mining, electroplating, metallurgical smelting, pesticide, and fertilizer use in agricultural fields [9].



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Furthermore, HMs that find their way into the environment linger and seriously endanger creatures that come into proximity to them due to toxicity. Although very low quality is crucial for the biological operations of plants and animals, but in high dose is toxic and inhibit metabolism in other organisms [10]. Toxic heavy metals, such as lead (Pb), cadmium (Cd), mercury (Hg), chromium (Cr), zinc (Zn), uranium (Ur), selenium (Se), silver (Ag), gold (Au), nickel (Ni), and arsenic (As), which are not useful to plants, inhibit plant growth, photosynthetic and enzymatic activities, and essential plant nutrition [11]. Moreover, even at low concentrations, heavy metals are carcinogenic to humans [12]. Conversely, bioremediators are biological agents used in bioremediation to help to clean contaminated sites. Among the most often used bioremediators are bacteria, archaea, and fungi [13]. The application of bioremediation, a biotechnological method that uses microorganisms to solve and eliminate environmental concerns caused by different pollutants through biodegradation.

Scientists are coming to an agreement on how to reduce pollutant release and mitigate their impacts using living creatures such as plants, which is known as phytoremediation or bacteria, which is referred to as bioremediation [14]. To address these issues, biological methods such as biosorption, bioaccumulation, biodegradation, and bioremediation are used to remove heavy metal ions, providing an appealing alternative to physicochemical methods [15]. These methods are potentially simple, low cost, more effective, ecofriendly, and a self-sustaining option for wastewater amelioration, which is gaining new attention nowadays [16]. Also, the chemical methods for heavy metal remediation pose health hazards such as toxicity and environmental risks, coupled with limitations like high costs and inefficiency under varying conditions. In contrast, microorganism-assisted approaches offer eco-friendly, sustainable, and cost-effective alternatives, leveraging natural mechanisms for safe and effective HMs removal [2,17].

The goal of this review is to explore current trends in the application/role of microorganisms in bioremediation with their interactions for nano-bioremediation, find the necessary background information to fill in any gaps in this theme area. In this study, the nono-bioremediation strategy provides a sustainable way for removing or minimizing HMs from the plants, soils and environments.

SOURCES OF HEAVY METALS

Natural resources, such as air, soil, and water ecosystems, have been discovered to contain HMs, agricultural chemicals, industrial solvents (especially chlorinated solvents), and other types of pollution [18]. The use of HMs by industry and agricultural sectors has resulted in massive amounts of HMs being relinquished and disposed of inadvertently in most ecosystems [19]. Several agricultural practices, such as the irreversible use of urban sewage sludge, industry-based practices, such as composting and burning of garbage in a variety of techniques, and vehicle emissions unintentionally introduce toxic metals (Cd, Cr, Pb, Hg, As, Cu, Zn, and Ni) into soils (Figure 1). Motor vehicle emissions (Pb) [20], engine wear (Cd, Cu, and Ni) [21], and tire abrasion (e.g., Zn). According to Hunter [22], arsenic could have positive effects on gene silencing and methionine metabolism in animals. Arsenic and cadmium are naturally present in very small amounts in the Earth's crust and probably weren't conscripted during evolutionary processes because they are less abundant than P and Zn, which occupy adjacent columns in the periodic table, respectively [23].

The accumulation of potentially harmful quantities of As and Cd in soils is largely due to anthropogenic activity [24]. Metal toxicity is critical for living organisms, including microbes, plants, animals, and humans. However, the toxicity varies depending on the

organism, while most of the 80 metals detected are essential for human functioning biology (e.g., Fe, Mg, Zn), others, such as Pb, Hg, and Cd, are among the oldest human toxicants [25]. A few heavy metals, including Fe, Cu, and Zn, are necessary microelements, whereas others, like Cd and Pb, have no beneficial function and are harmful even at low quantities (Figure 1). The contamination of soil and aquatic ecosystems is important, because metals are not biodegradable like other organic contaminants, they accumulate in terrestrial, aquatic, and marine ecosystems [26].

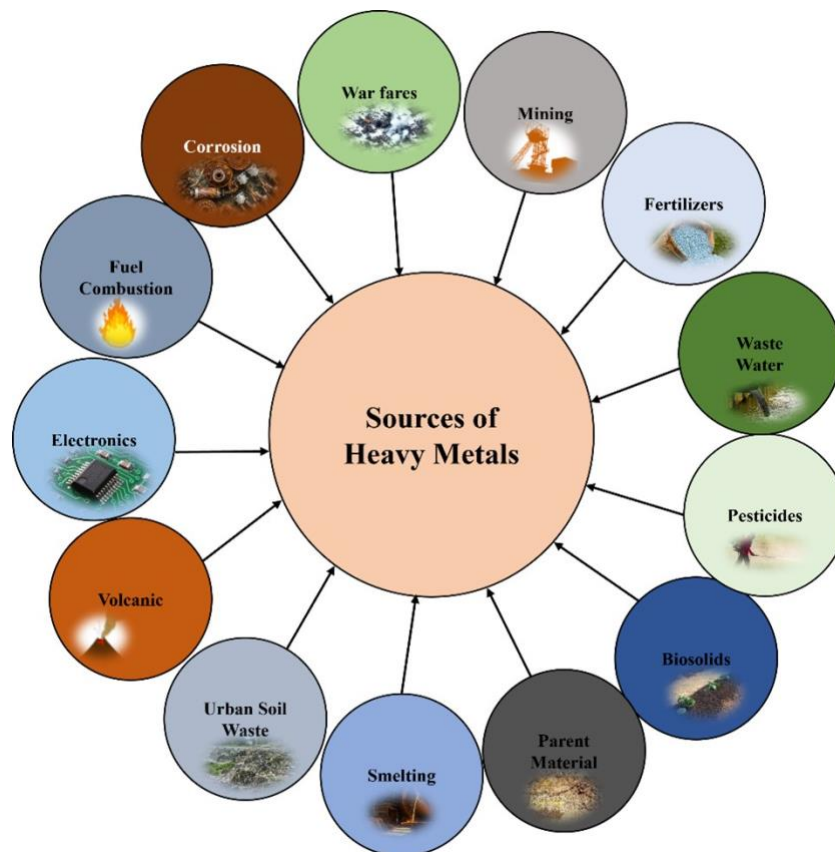


Figure 1. Sources of HMs in the environment. The figure illustrates the various anthropogenic and natural sources of heavy metals, such as industrial discharge, agricultural runoff, natural mineral deposits, soil waste and toxic chemicals.

HEAVY METALS CONTAMINATION AND PUBLIC HEALTH RISK

HMs contamination problem is one of the major concerns worldwide. HMs toxicity varies due to their toxicity levels. The HMs are fluently added to environments from different sources (Figure 1), and those can easily deposit into living organisms [27]. They are all metals that have an atomic weight higher than that of iron (55.8 gmol^{-1}), and they are found in the environment. However, some metals have an atomic weight lower than that of Fe. For example, Cr, and some metalloids, like As and Se, are also called "heavy metals" [28]. Many types of HMs can be micronutrients for humans. These include Cu, Fe, Mn, Mo, Zn, and Ni. They can also be toxic to humans if being exposed to them for a longer period, like Hg, Pb, Cd, Cu, Ni, and Co. Contamination by HMs has a lot of bad effects, not just on animals and plants but also on human health [29]. As an example, Zn is a component of a few enzymes, such as enzymes that break down carbohydrates, proteins, and peptides, as well as enzymes that make RNA and ribosomes in plants. Copper helps plants do a lot of things, like photosynthesis, respiration, carbohydrate distribution, nitrogen and cell wall metabolism, seed production, and disease resistance, but at high concentrations, these metals can harm cells [30]. The Cd is highly toxic for

biological processes and very harmful to organisms when it builds up even in low quantities. Deposition of HMs depends on metal ions specific ion-binding processes to specific locations, and cellular structure variability. HMs ions, have a strong electrostatic attraction and great binding affinities with these same locations. The toxicity outputs of HMs are critical for basic genetic molecules of organisms. The structures and biomolecules, such as cell wall enzymes, DNA, and RNA, become unstable because of this, which is ultimately responsible for occurring mutations at molecular levels, as a result altering genetic levels, physiological difficulties, illnesses, and even cancer [31].

HEAVY METAL TOXICITY IN PLANTS, SOIL, AND ENVIRONMENT

Globally, due to the persistence, high toxicity, and recalcitrant nature, metal contamination has now become a serious concern in plants, soil, and the environment (Figure 2). These toxic metals have posed a serious threat to the environmental stability and health of all living organisms [2]. Like other living organisms, plants are not resistant to high concentrations of HMs in the air due to human activities and environment. Trace amounts of HMs enhance plants by acting as essential micronutrients. According to Lopez-Vargas et al. [32], copper can improve the flavor and color of floral arrangements, fruits, and vegetables by increasing sugar content in plants. Zn is an essential component of the enzymatic system as well as the metabolic processes of plants [33], and photosynthetic compounds are particularly influenced by HMs [34]. Heavy metals have been shown to accumulate in plants, where they interfere with the normal metabolic and biological processes in the plant, and eventually leading to severe yield losses [35].

HMs prevent seed germination by adversely influencing the processes, which in turn reduces the establishment of the entire stand [36]. Due to the high concentration of malondialdehyde (MDA) and H₂O₂, HM also impairs the water status of plants, damages the stability of their membranes, and increases the loss of crucial osmolytes. It shows that different chemical, physical, and biological methods for HMs removal from soils have been in practice globally. Plants consume and accumulate HM that is present in the soil at very high concentrations and eventually reaches human nutrition through the food chain [37]. Recently, microbes have gained a lot of attention from scientists worldwide. Biosorption, bioaccumulation, biovolatilization, biomineralization, oxidation and reduction, bioleaching, and the synthesis of bio-surfactants are some of the methods by which the bacteria extract the heavy metals from the soil [17].

Plants and microorganisms are used as biological methods to treat HMs containing polluted soils [38]. Hints, there are also certain drawbacks to these techniques in terms of long durations, environmental sensitivity, and toxicity of contaminants [39]. High levels of HMs in soil and water are representative examples of human activities which have a significant effect on the environment and present a huge risk [40,41]. According to Abd Elnabi et al. [42], living plants and animals are at serious risk due to the toxic HM's persistence in the soil environment. For terrestrial plants, the primary points of contact with hazardous heavy metals (HMs) are the roots [43]. Additionally, microbes lower the concentration of heavy metals (HMs) in soil. For example, *Aspergillus niger* shown a notable capacity to bioaccumulate Cd and Cr [44], and *Stenotrophomonas rhizophila* also considerably reduced Pb and Cu by 76.9% and 83.4%, respectively [45]. Microbes have a large surface area to adsorb the HMs because of their small size, which lowers the total amount of HMs available [46]. Soil biology is indispensable with respect to soil quality maintenance, which again is very important for sustainable agriculture. Human activities have emerged as the prime source of HMs and have disturbed the soil microbes, soil fertility, and productivity [47].

However, their bioaccumulation and biomagnification attributes in the food chain are highly threatening to the environment due to their accumulation in soil and plants [48]. Application of chemical fertilizers and pesticides can enhance the risk of HMs contamination in the soil. The chemical toxicity outputs lead to build-up in crop tissue grown in the contaminated soil [49]. Microbial bioremediation methods have recently demonstrated significant promise in cleaning up contaminated soils. One type of green technology is the utilization of microorganisms' metabolic processes to remove heavy metal contamination, and most HMs have been classified as hazardous overall.

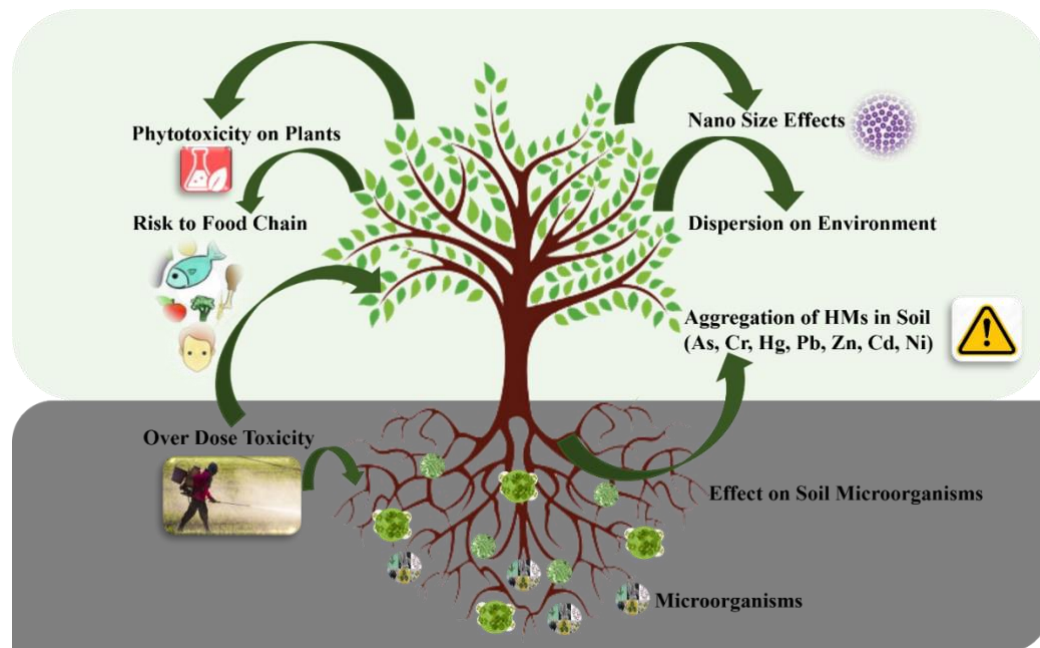


Figure 2. Toxicity of HMs in plants, soil, and environment. The figure represents the toxic effects of HMs on plant physiology, soil fertility, and overall environmental health. The figure highlights specific mechanisms of toxicity such as disruption of enzymatic activities and oxidative stress in plants, alongside soil degradation.

NANO-BIOREMEDIATION OF HEAVY METALS USING MICROORGANISMS

Nanoparticles-based bioremediation is an emerging and highly efficient approach for large-scale environmental cleanup, minimizing toxic repercussions. In the relentless march of technology, bioremediation has evolved into nano-bioremediation, employing nanoparticles and microbes to offer eco-friendly solutions for tackling hazardous environmental pollutants (Figure 3).

Microorganisms-based HMs removing or minimizing are more efficient approaches compared to traditional methods [50]. Metals like As, Cd, and Pb are highly toxic even at low concentrations [51]. Microbial bioremediation immobilizes these metals; for example, *Morganella psychrotolerans* produces silver nanoparticles for heavy metal removal [52], while iron oxide nanoparticles with polyvinyl pyrrolidone (PVP) and *Halomonas* sp. effectively remediate Pb and Cd [53]. Additionally, silica nanoparticles, *Pseudomonas aeruginosa*, and graphene oxide remove polycyclic aromatic hydrocarbons (PAHs) [54], and *Halomonas* immobilized with magnetic NPs degraded Pd metal [55]. Moreover, Bacteria synthesize diverse nanoparticles used for immobilization and mobilization of metals [56], with strains like *Bacillus cereus* (PMBL-3) and *Lactobacillus macroides* (PMBL-7) effectively remediating HMs such as Cd, Cr, Pd, and Cu [57]. Interestingly, Myconanotechnology employs fungi for bioremediation (Table 1), with various mushroom species effectively remediating soil contaminants [58, 59]. Fungi such

as *Fusarium solani* are increasingly being used in the process of nanoparticle manufacturing, because of their resistance to heavy metals [60], while *Trichoderma harzianum* degrades pentachlorophenol, and *Cryptococcus* sp. displays resilience to HMs [61].

The chemical structure and molecular weight of microplastics (MPs), along with environmental conditions, influence microbial-driven degradation. This process involves biodeterioration, bio-fragmentation, biosynthesis, and mineralization [82]. Although the precise mechanism of *Pseudomonas* sp. in degrading MPs particles remains elusive, research suggests the involvement of chitinase degradation [83]. Microorganisms utilize metal-based nanoparticles (MNPs) as a carbon source for growth, aiding in the degradation of high molecular-weight plastics [84]. The microbial degradation of plastic fragments offers a green solution, but controlling changes in plastic pollution relies on various factors. Therefore, the utilization of effective microbes can be a suitable approach for eliminating MNPs [81, 85]. However, microbial degradation of MNPs is still in its infancy, as the characterization process is very slow, and incomplete mineralization is another limitation [86]. MNPs present in municipal solid waste (MSW) can harbor various microinorganic and organic pollutants, posing risks to the environment and human health as they enter the food chain [87]. Indigenous microbial communities within MSW and sewage sludge exhibit plastic degradation capabilities. Recent studies have shown that mesophilic *Stenotrophomonas panacihumi* can convert polypropylene (PP) into low and high-molecular-weight forms after 90 days [88]. The persistence of antibacterial nanoparticles beyond a threshold poses a threat to soil microbes, potentially inhibiting nitrogen-fixing microbes, leading to stunted plant growth and reduced production [89].

In the perspective of environmental bioremediation, leveraging nanotechnology alongside microbial assistance emerges as a highly efficient strategy, as evidenced by the effective removal of HMs and pollutants from soil and wastewater. While nano-bioremediation offers promising solutions, the persistence of antibacterial nanoparticles poses challenges to soil microbial communities and ecosystems. Despite advancements, microbial degradation of micro-nano plastics remains in its nascent stages, highlighting the need for further research to optimize biodegradation processes and minimize environmental risks.



Figure 3. Role of nanoparticles in environmental clean-up. The figure depicts the sources of nanoparticles and their mechanisms of action in remediation of contaminated soil, water, and air/environment. It outlines the enzymatic processes involved and the effectiveness of various nanoparticles in sequestering or degrading heavy metal pollutants. These extended captions provide a comprehensive description of each figure, ensuring that readers can fully understand the implications and contexts of the visual data presented.

Table 1. Potential implication of microorganisms for bioremediation of HMs from plants, soils and environments.

S. No.	Microorganisms	Heavy metals	Remarks	Ref.
1	<i>Azospirillum brasilense</i> , <i>Bradyrhizobium japonicum</i>	As	Mortality reduction, increase plant growth, and nitrogen content in <i>Glycine max.</i>	[62]
2	<i>Bacillus cereus</i>	Cd, Cu, Ni, Pb, Zn	Improves phytoremediation efficiency in <i>Zea mays.</i>	[63]
3	<i>Pseudomonas lurida</i>	Cu	Increase Cu accumulation in roots and leaves of <i>Helianthus annuus.</i>	[64]
4	<i>Aspergillus niger</i> , <i>Ascophyllum nodosum</i> , <i>Bacillus firmus</i> , <i>Chlorella fusca</i> , <i>Oscillatoria angustissima</i>	Pb, Zn, Cd, Cr, Cu, Ni	Removal of HMs from wastewater	[65]
5	<i>Bacillus subtilis</i> , <i>B. licheniformis</i> , <i>Streptomyces pactum</i>	Cu, Cd, Pb, Zn	Increase growth of <i>Brassica juncea.</i>	[66]
6	<i>Micrococcus luteus</i>	Pb	Remediate moderately Pb from soil.	[67]
7	<i>Funnelliformis mosseae</i>	Cd	Enhance growth and Cd accumulation of <i>Solanum nigrum.</i>	[68]
8	<i>Bacillus</i> sp.	As	Enhance As uptake and removal capacity in <i>Vallisneria denseserrulata.</i>	[69]
9	<i>Glomus mosseae</i>	Cd, As, Pb	Enhance plant growth, photosynthetic pigments in <i>Pisum sativum.</i>	[70]
10	<i>Saccharomyces cerevisiae</i>	Cr, Ni, Cu, Zn	HMs removal using dead biomass	[71]
11	<i>Rhizophagus irregularis</i>	Cr	Enhance the photosynthetic performance, tolerance index, transportation index of <i>Brachiaria mutica.</i>	[72]
12	<i>Spirogyra</i> sp., <i>Cladophora</i> sp.	Pb, Cu	Wastewater treatment	[73]
13	<i>Spirogyra</i> sp., <i>Spirullina</i> sp.	Cr, Cu, Fe, Mn, Zn	HMs uptake	[74]
14	<i>Humicola</i> sp.	As	Enhance plant growth of <i>Bacopa monnieri.</i>	[75]
15	<i>Hydrodictyon</i> , <i>Oedogonium</i> , <i>Rhizoclonium</i> sp.	V, As	Biosorption dried algal biomass	[76]
16	<i>Rhizophagus</i> sp., <i>Funelliformis</i> sp.	As	Increase aerial parts of <i>Pteris vittata.</i>	[77]
17	<i>Piriformospora indica</i>	As	Accumulate As in roots of <i>Artemisia annua.</i>	[78]
18	<i>Rhizophagus irregularis</i>	Cd, Zn	Increase the activities of ascorbate peroxidase (APX) and SOD in <i>Phragmites australis.</i>	[79]
19	<i>Streptomyces</i> spp. B1, B2, B3	Cd, Pb, and Zn	Enhance plant biomass and decrease oxidative stress in <i>Salix dasyclados.</i>	[80]
20	<i>Mesorhizobium loti</i> , <i>Ensifer adhaerens</i> , <i>Rhizobium radiobacter</i>	Cd, Pb, Cr, Cu, Zn	Increase nodule number in <i>Robinia pseudoacacia.</i>	[81]

LIMITATIONS AND FUTURE PROSPECTS

The application of microbes in remediation of HMs is known as nano-bioremediation and has drawn significant attention due to its efficient and environmentally friendly nature. However, several limitations have also been noticed during its widespread application. Saharan et al. [90] observed, the HMs toxicity in microbial communities

affect the metabolic activity of these microorganisms and consequently their overall efficacy. In contrast, the environmental persistence of engineered NPs expands concerns about their potential ecotoxicological impacts, including bioaccumulation and non-target effects [91]. Laboratory findings often struggle to translate to field conditions due to variations in soil composition, pH, temperature, and competing ions, alongside unpredictable interactions among nanoparticles, plants, and microbes that are not fully understood [92, 93]. The characteristics of engineered nanoparticles can change due to agglomeration or dissolution, affecting their reactivity and potential toxicity [94]. Additionally, the long-term impacts of these nanoparticles, such as their accumulation in plant tissues or migration into water sources, remain poorly characterized. The scalability of nanoparticles is limited by expensive synthesis techniques and the high costs of large-scale field applications, especially in resource-constrained regions. Nanoparticles can introduce new environmental contaminants and potentially disrupt soil microbiomes, altering nutrient cycles and ecosystem balance due to high concentrations. The adoption of nanotechnology is complicated by inconsistent regulations across different regions and the absence of standardized protocols for assessing environmental safety and long-term impacts [95].

The environmentally friendly biodegradable nanoparticles will alleviate the worries about persistence and eco-toxicity effects [68]. Integration with omics technologies, particularly genomics and proteomics, will provide deeper insight into the interaction of microbes with nanoparticles, thereby facilitating the design of tailored bioremediation strategies for specific contaminants [96]. Real-time monitoring systems are integrated with nano-bioremediation, also including biosensors that enhance process efficiency while ensuring site-specific applicability [17]. Advancing the field of nano-bioremediation of heavy metals through plant-microorganism interactions involves several promising directions. Elucidating mechanisms at the molecular level using comprehensive omics approaches and real-time monitoring techniques, such as synchrotron-based spectroscopy, can support deep insights into processes involved in HM and nanoparticle transformations.

Developing eco-friendly and cost-effective nanomaterials through green synthesis methods and creating biodegradable nanoparticles can minimize environmental impact and production costs. Integrating nano-bioremediation with different remediation strategies, such as phytoremediation and chemical techniques, can enhance contaminant removal efficiency and manage complex pollution scenarios more effectively. Conducting field trials and validating long-term effectiveness under various conditions is crucial for assessing the feasibility and sustainability of nano-bioremediation methods. Additionally, the application of advanced models and tools, including AI and machine learning, can optimize remediation processes and customize approaches based on local environmental conditions. Harmonization and the establishment of standardized risk-benefit analyses are essential for gaining broader acceptance and ensuring the responsible implementation of nano-bioremediation technologies. These efforts will help refine nano-bioremediation strategies, making them more efficient, scalable, and environmentally safe. Therefore, the application of advanced nano-based tools and sustainable restoration of ecosystems from HMs-contaminated environments would be supportive for making smart and HMs-free green environments.

CONCLUSIONS

This study explores effective strategies for microorganism-assisted HMs remediation, the role of plant-microbe interactions for nano-bioremediation of HMs, and strategies for transforming toxic HMs into less toxic forms. This study updates the techniques like biomineralization, biostimulation, and mycoremediation, discusses how microbe-assisted phytoremediation is crucial for minimizing HMs contamination, and shows sustainable solutions for maintaining of HMs toxicity. This review also highlights the prospect of using nano-environmental biotechnology tools for mitigating HMs toxicity in plants, soils, and environments. This updated study with eco-friendly approaches to HMs removal from environments could be useful for minimizing HMs toxicity and converting to smart-green earth.

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AUTHOR CONTRIBUTIONS

KD conceived the research plan, and supervised the study. KD, MAAM, and AS wrote the initial draft of the manuscript. KD, FMA, and SD edited the manuscript. All authors have approved the final version.

CONFLICTS OF INTEREST

There is no conflict of interest among the authors.

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